METHODS FOR ASSOCIATING OR DISSOCIATING GUEST MATERIALS WITH A METAL ORGANIC FRAMEWORK, SYSTEMS FOR ASSOCIATING OR DISSOCIATING GUEST MATERIALS WITHIN A SERIES OF METAL ORGANIC FRAMEWORKS, AND GAS SEPARATION ASSEMBLIES

CROSS REFERENCE TO RELATED APPLICATION

This application claims priority to U.S. Provisional Patent Application Ser. No. 61/320,445 which was filed on Apr. 2, 2010, the entirety of which is incorporated by reference herein.

TECHNICAL FIELD

The present disclosure relates to the use of metal organic frameworks.

BACKGROUND

Recently, metal organic frameworks have been proposed for use in various capacities. These capacities include but are not limited to the separation of molecules or materials from mixtures that include the molecules or materials. As an example, in various applications, metal organic frameworks have been proposed for use as materials that can be used to separate carbon dioxide from methane, for example.

In accordance with other applications, metal organic frameworks have also been utilized to retain certain molecules in higher density than they would be retained at when super pressurized. As an example, metal organic frameworks have been proposed for use as hydrogen storage tanks.

In these applications, in the past, the metal organic frameworks have been configured to selectively adsorb or desorb or associate or dissociate certain materials. As an example, the temperature and/or pressure of the metal organic framework can be manipulated, as well as the chemical and/or geometric structure of the metal organic framework, to facilitate either the association or adsorption, or the dissociation or desorption of the specific materials.

The present disclosure provides methods for using metal organic frameworks as well as systems that include metal organic frameworks and assemblies that include metal organic frameworks.

BRIEF DESCRIPTION OF THE DRAWINGS

Embodiments of the disclosure are described below with reference to the following accompanying drawings.

FIG. **1** is a configuration of a metal organic framework according to an embodiment of the disclosure.

FIG. **2** represents configurations of metal organic frameworks according to an embodiment of the disclosure.

FIG. **3** represents configurations of metal organic framework and mixtures that include guest materials depicted according to an embodiment of the disclosure.

FIG. **4** is a synthetic scheme of a metal organic framework according to an embodiment of the disclosure.

FIG. **5** represents a configuration of metal organic framework according to an embodiment of the disclosure.

FIG. **6** represents a system including metal organic frameworks according to an embodiment of the disclosure.

FIG. **7** represents a system including metal organic framework according to an embodiment of the disclosure.

FIG. **8** represents a system including metal organic framework according to an embodiment of the disclosure.

FIG. **9** represents electrolyte structures according to an embodiment of the disclosure.

FIG. **10** represents a system including metal organic framework according to an embodiment of the disclosure.

FIG. **11** represents a system including metal organic framework according to an embodiment of the disclosure.

DESCRIPTION

This disclosure is submitted in furtherance of the constitutional purposes of the U.S. Patent Laws "to promote the progress of science and useful arts" (Article 1, Section 8).

Methods for releasing associated guest materials from a metal organic framework are provided with example methods including altering the oxidation state of at least a portion of the metal of the metal organic framework to dissociate at least a portion of the guest materials from the framework. Example methods for associating guest materials with a metal organic framework are also provided with example methods including altering the oxidation state of at least a portion of the metal of the metal organic framework to associate at least a portion of the guest materials with the framework.

Methods are provided for selectively associating or dissociating guest materials with a metal organic framework. Example methods can include altering the oxidation state of at least a portion of the metal of the metal organic framework to associate or dissociate at least a portion of the guest materials with the framework.

Systems for associating or dissociating guest materials within a series of metal organic frameworks are provided. Example systems can include at least two individual metal organic frameworks, with one of the individual metal organic frameworks configured to dissociate guest materials, and the other configured to associate guest materials. One framework can include at least some metals of one oxidation state and the other framework can include the same metals of another oxidation state.

Gas separation assemblies are provided. Example assemblies can include a plurality of individual cells housing metal organic framework, with the plurality comprising at least one series of cells and another series of cells. Each series can include at least two electrically isolated individual cells. The assemblies can also include a channel between the one series of cells and the other series of cells. Power sources and a controller coupled to both the assembly and the power source can also be included in certain embodiments.

The methods, systems, and assemblies of the present disclosure will be described with reference to FIGS. **1-11**. Referring first to FIG. **1**, a metal organic framework configuration **10** is shown that includes metal organic framework **12** conductively coupled via contact **16** and conductive conduit **18** to power source **20**. Framework **12** can include metals coupled to organic components. Framework **12** may be configured to define open sites designed to receive guest materials. The open sites may be defined by more than one metal organic complex, for example. At least a portion of the metals of the metal organic framework should be electrically responsive. More than one metal may be included in metal organic complex **13** having organic portion **14** and metal portion **15**.

Metal portion **15** of complex **13** can include metals and, according to example implementations, the oxidation state of at least some of the metals will change upon application of differing voltages to the framework. The metals can include transition state metals. In accordance with example implementations, upon changing the oxidation state of some of the