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(54) **ENERGY STORAGE DEVICES HAVING ANODES CONTAINING MG AND ELECTROLYTES UTILIZED THEREIN**

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(58) **Field of Classification Search**
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See application file for complete search history.

(56) **References Cited**

U.S. PATENT DOCUMENTS

6,713,212	B2	3/2004	Aurbach et al.	
2008/0182176	A1	7/2008	Aurbach et al.	
2008/0226983	A1 *	9/2008	Odani et al.	429/200
2013/0108919	A1 *	5/2013	Matsui	429/188
2014/0154592	A1 *	6/2014	Mohtadi et al.	429/344

OTHER PUBLICATIONS

Kim, DJ, et al., "Electrochemical properties of magnesium electrolyte with organic solvent," Nov. 23, 2007, pp. 70-73, IOP Publishing, Physica Scripta, Korea.

Aurbach, D, et al., "Electrolyte Solutions for Rechargeable Magnesium Batteries Based on Organomagnesium Chloroaluminate Complexes," Dec. 20, 2001, pp. A115-A121, vol. 149, Journal of Electrochemical Society, Israel.

Mohtadi, R, et al., "Magnesium Borohydride: From Hydrogen Storage to Magnesium Battery," Aug. 21, 2012, pp. 9780-9783, vol. 51, Angewandte Communications, Wiley, Online.

* cited by examiner

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(57) **ABSTRACT**

For a metal anode in a battery, the capacity fade is a significant consideration. In energy storage devices having an anode that includes Mg, the cycling stability can be improved by an electrolyte having a first salt, a second salt, and an organic solvent. Examples of the organic solvent include diglyme, triglyme, tetraglyme, or a combination thereof. The first salt can have a magnesium cation and be substantially soluble in the organic solvent. The second salt can enhance the solubility of the first salt and can have a magnesium cation or a lithium cation. The first salt, the second salt, or both have a BH₄ anion.

16 Claims, 4 Drawing Sheets