

1

LOW FIELD MOBILITY SEPARATION OF IONS USING SEGMENTED CYLINDRICAL FAIMS

This application claims benefit from U.S. Provisional Application No. 60/598,838 filed Aug. 05, 2004, the entire contents of which are incorporated herein by reference.

FIELD OF THE INVENTION

The instant invention relates generally to High Field Asymmetric Waveform Ion Mobility Spectrometry (FAIMS). In particular, the instant invention relates to methods and apparatus for high-resolution separation of ions based on their high-field and low-field mobility properties.

BACKGROUND

High sensitivity and amenability to miniaturization for field-portable applications have helped to make ion mobility spectrometry (IMS) an important technique for the detection of many compounds, including narcotics, explosives, and chemical warfare agents as described, for example, by G. Eiceman and Z. Karpas in their book entitled "Ion Mobility Spectrometry" (CRC, Boca Raton, 1994), the contents of which are incorporated herein by reference. In IMS, gas-phase ion mobilities are determined using a drift tube with a constant electric field. Ions are separated in the drift tube on the basis of differences in their drift velocities. At low electric field strength, for example 200 V/cm at approximately ambient atmospheric pressure, the drift velocity of an ion is proportional to the applied electric field strength, and the mobility, K , which is determined from experimentation, is independent of the applied electric field. Additionally, in IMS the ions travel through a bath gas that is at sufficiently high pressure that the ions rapidly reach constant velocity when driven by the force of an electric field that is constant both in time and location.

E. A. Mason and E. W. McDaniel in their book entitled "Transport Properties of Ions in Gases" (Wiley, N.Y. 1988), the contents of which are incorporated herein by reference, teach that at high electric field strength, for instance fields stronger than approximately 5,000 V/cm, the ion drift velocity is no longer directly proportional to the applied electric field, and K is better represented by K_H , a non-constant high field mobility term. The dependence of K_H on the applied electric field has been the basis for the development of high field asymmetric waveform ion mobility spectrometry (FAIMS). Ions are separated in FAIMS on the basis of a difference in the mobility of an ion at high field strength, K_H , relative to the mobility of the ion at low field strength, K . In other words, the ions are separated due to the compound dependent behavior of K_H as a function of the applied electric field strength.

In general, a device for separating ions according to the FAIMS principle has an analyzer region that is defined by a space between first and second spaced-apart electrodes. The first electrode is maintained at a selected dc voltage, often at ground potential, while the second electrode has an asymmetric waveform $V(t)$ applied to it. The asymmetric waveform $V(t)$ is composed of a repeating pattern including a high voltage component, V_H , lasting for a short period of time t_H and a lower voltage component, V_L , of opposite polarity, lasting a longer period of time t_L . The waveform is synthesized such that the integrated voltage-time product, and thus the field-time product, applied to the second electrode during each complete cycle of the waveform is

2

zero, for instance $V_H t_H + V_L t_L = 0$; for example +2000 V for 10 μ s followed by -1000 V for 20 μ s. The peak voltage during the shorter, high voltage portion of the waveform is called the "dispersion voltage" or DV, which is identically referred to as the applied asymmetric waveform voltage.

Generally, the ions that are to be separated are entrained in a stream of gas flowing through the FAIMS analyzer region, for example between a pair of horizontally oriented, spaced-apart electrodes. Accordingly, the net motion of an ion within the analyzer region is the sum of a horizontal x-axis component due to the stream of gas and a transverse y-axis component due to the applied electric field. During the high voltage portion of the waveform, an ion moves with a y-axis velocity component given by $v_H = K_H E_H$, where E_H is the applied field, and K_H is the high field ion mobility under operating electric field, pressure and temperature conditions. The distance traveled by the ion during the high voltage portion of the waveform is given by $d_H = v_H t_H = K_H E_H t_H$, where t_H is the time period of the applied high voltage. During the longer duration, opposite polarity, low voltage portion of the asymmetric waveform, the y-axis velocity component of the ion is $v_L = K E_L$, where K is the low field ion mobility under operating pressure and temperature conditions. The distance traveled is $d_L = v_L t_L = K E_L t_L$. Since the asymmetric waveform ensures that $(V_H t_H) + (V_L t_L) = 0$, the field-time products $E_H t_H$ and $E_L t_L$ are equal in magnitude. Thus, if K_H and K are identical, d_H and d_L are equal, and the ion is returned to its original position along the y-axis during the negative cycle of the waveform. If at E_H the mobility $K_H > K$, the ion experiences a net displacement from its original position relative to the y-axis. For example, if a positive ion travels farther during the positive portion of the waveform, for instance $d_H > d_L$, then the ion migrates away from the second electrode and eventually is neutralized at the first electrode.

In order to reverse the transverse drift of the positive ion in the above example, a constant negative dc voltage is applied to the second electrode. The difference between the dc voltage that is applied to the first electrode and the dc voltage that is applied to the second electrode is called the "compensation voltage" (CV). The CV prevents the ion from migrating toward either the second or the first electrode. If ions derived from two compounds respond differently to the applied high strength electric fields, the ratio of K_H to K may be different for each compound. Consequently, the magnitude of the CV that is necessary to prevent the drift of the ion toward either electrode is also different for each compound. Thus, when a mixture including several species of ions, each with a unique K_H/K ratio, is being analyzed by FAIMS, only one species of ion is selectively transmitted to a detector for a given combination of CV and DV. In one type of FAIMS experiment, the applied CV is scanned with time, for instance the CV is slowly ramped or optionally the CV is stepped from one voltage to a next voltage, and a resulting intensity of transmitted ions is measured. In this way a CV spectrum showing the total ion current as a function of CV, is obtained.

Numerous ionization sources, including atmospheric pressure ionization sources, have been described for use with FAIMS. Non-limiting examples include electrospray ionization and variants thereof, thermospray, atmospheric pressure chemical ionization, corona discharge, radioactive sources (including ^{63}Ni and other nuclides). As some non-limiting examples, ions are detected using electrometers for measuring electric current from discharge of the ions, detected by measurement of currents induced by the presence of the ions, detected indirectly using chemical reac-